

107 E (a)

$$T(^{\circ}\text{F}) = 1.8 T(^{\circ}\text{C}) + 32$$

$$V \left(\frac{\text{miles}}{\text{h}} \right) = 1.609 \frac{\text{km}}{\text{h}}$$

The coefficient 0.0203 has units of $\frac{\text{hr}}{\text{mi}}$

\therefore it changes to $\frac{0.0203}{1.609} = 0.0126$

and the coefficient 0.304 changes to

$$\frac{0.304}{1.609} = 0.240$$

$$T(^{\circ}\text{F}) = 91.4 - [91.4 - T_{\text{amb}}] [0.475 - 0.0203V + 0.304 \sqrt{V/1.609}]$$

$$\Rightarrow 33.0 - (33.0 - T_{\text{amb}}^{\circ}\text{C}) (0.475 - 0.0126V + 0.240\sqrt{V})$$

(b)

$$1 \frac{\text{km}}{\text{h}} = 0.28 \text{ m/s}$$

$$33.0 - (33.0 - T_{\text{amb}}^{\circ}\text{C}) (0.475 - 0.04536V + 0.84\sqrt{V})$$

110

$$D = 10 \text{ m}$$

$$V = \frac{4}{3} \pi R^3 = \frac{1}{6} \pi D^3 = \frac{1}{6} \cdot 3.14 \cdot 1000 \text{ (m}^3\text{)} = 523.6 \text{ m}^3$$

$$F_B = \rho_{\text{air}} g V_{\text{balloon}}$$

$$= (1.16 \frac{\text{kg}}{\text{m}^3}) (9.81 \text{ m/s}^2) (523.6 \text{ m}^3)$$

$$= 5958.4 \frac{\text{kg} \cdot \text{m}}{\text{s}^2} = 5958.4 \text{ N} \quad (\because 1 \frac{\text{kg} \cdot \text{m}}{\text{s}^2} = 1 \text{ N})$$

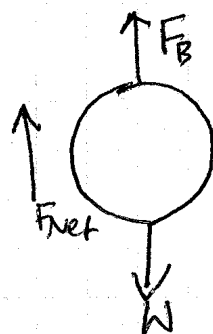
$$M = m_{\text{balloon}} + m_{\text{people}} = \rho_{\text{He}} V_{\text{balloon}} + 2 \times 70 \text{ kg}$$

Pset #2

(2)

$$\rho_{He} = \frac{1}{7} \rho_{air} = \frac{1.16}{7} \text{ kg/m}^3 = 0.166 \text{ kg/m}^3$$

$$M = (0.166) (523.6 \text{ (kg)}) + 140 \text{ kg} \\ = \underline{226.8 \text{ kg}}$$



$$F_{net} = M a \Rightarrow a = \frac{F}{M} \text{ (m/s}^2\text{)} =$$

$$F_{net} = 5928 - (226.8)(9.81) = 2225 \text{ N}$$

$$a = \frac{2225}{226.8} = 9.81 \text{ m/s}^2$$

118) $P_{atm} = 101.325 (1 - 0.022562)^{5.256}$

Simple Substitution:

Atlanta	97.7 kPa
Denver	83.4 kPa
M. city	76.5 kPa
Mt. Ev.	31.4 kPa

119) $\rho_{Liq} = 0.81 \text{ kg/L} = 0.81 \times 1000 \frac{\text{kg}}{\text{m}^3} = 810 \frac{\text{kg}}{\text{m}^3}$

$$P_{gage} = P_{abs} - P_{atm}$$

$$= \rho g h$$

$$= (810)(9.81)(0.08)$$

$$= \underline{636 \text{ Pa}}$$

$$L = \frac{h}{\sin \theta} = 13.9 \text{ cm}$$

123)

$$T = T_0 - \beta z$$

$$T_0 = 288.15 \text{ K}$$

$$\beta = 0.0065 \text{ K/m}$$

$$\left(\frac{\text{kg} \cdot \text{m}}{\text{s}^2} \right) \left(\frac{1}{\text{m}^2} \right) = \text{Pa}$$

Pset #2

$$g(z) = \frac{g_0}{(1 + z/6,370,320)^2}$$

$$g_0 = 9.807 \text{ m/s}^2 \quad (3)$$

(a) $g = g_0$

$$dP = -\rho g dz$$

$$\rho = \frac{P}{RT} = \frac{P}{R(T_0 - \beta z)}$$

$$dP = - \frac{P}{R(T_0 - \beta z)} \cdot g dz$$

$$\int_{P_0}^P \frac{dP}{P} = - \int_0^z \frac{g dz}{(T_0 - \beta z)} = \frac{g}{\beta} \ln\left(\frac{T_0 - \beta z}{T_0}\right) = \ln P/P_0$$

$$C = \frac{g}{R} \quad \frac{P}{P_0} = \left(\frac{T_0 - \beta z}{T_0}\right)^{g/\beta R}$$

$$P = P_0 \left[1 - \frac{\beta z}{T_0}\right]^{g/\beta R}$$

(b) $dP = - \frac{P}{R(T_0 - \beta z)} \frac{g_0}{(1 + z/c_1)^2} dz$

$$\int_{P_0}^P \frac{dP}{P} = - \int_0^z \frac{g_0}{R(T_0 - \beta z) (1 + z/c_1)^2} dz$$

Two approaches. Use Integral table for exact answer.

(2) Use Numerical Integration (easier) if z is given and generate a table for z .